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3D-QSAR (Quantitative Structure–Activity Relationships) Studies on Urea Derivatives as Inhibitors of p38 MAP Kinase

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Abstract

Motivation. Inhibition of the p38 MAP kinase pathway has been shown to be beneficial in the treatment of inflammatory diseases. A three-dimensional quantitative structure–activity relationship (3D-QSAR) model, based on molecular field analysis (MFA), was developed for 41 urea derivatives that are inhibitors of p38 mitogen-activated protein kinase (p38 MAPK). The 3D-QSAR model was developed using 32 compounds and its predictive ability was assessed using a test set of 9 compounds.

Method. MFA studies were performed using the QSAR module of Cerius².

Results. The predictive 3D-QSAR model has a conventional $r^2 = 0.901$ and a cross-validated coefficient $r^2_{cv} = 0.799$. The model was able to efficiently predict the activities of the compounds of the test set ($r^2_{pred} = 0.740$), suggesting that it can be used for the development of new p38 MAPK inhibitor candidates, useful for treatment of chronic inflammatory states.

Conclusions. The structural features indicated by this 3D-QSAR analysis provide a helpful guideline to design novel compounds with inhibitory activities against p38 MAPK.

Keywords. p38 MAP kinase, QSAR, MFA, urea derivatives.

Abbreviations and notations

MFA, Molecular Field Analysis

TNF, Tumor necrosis factor

SAR, Structure–Activity Relationship

r^2_{cv} , cross-validated r^2

PRESS, predicted sum of squared residuals

r^2_{pred} , predictive r^2

QSAR, quantitative structure–activity relationships

IL, Interleukin

MCS, minimum common subgraph

R^2 , conventional r^2

R^2_{bs} , bootstrap r^2

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1 INTRODUCTION

p38 MAP kinase is a member of Ser/Thr kinase family [1] of the mitogen–activated protein (MAP) kinase super family [2,3,4,5]. p38 MAP kinase plays a very important part in diseases, such as asthma, osteoarthritis and rheumatoid arthritis, chronic inflammatory autoimmune diseases. Therefore, the inhibition of p38 MAP kinase would potentially prevent the underlying pathophysiology in the inflammatory diseases, which makes p38 MAP kinase an attractive target for drug discovery [6,7]. The pyridinylimidazole compounds, exemplified by SB203580 (Figure 1), were originally developed as inflammatory cytokine synthesis inhibitors, and were later identified as selective inhibitors of p38 MAP kinase. Thereafter, structurally diverse p38 MAP kinase inhibitors have been synthesized and tested extensively to seek potential therapy for inflammatory diseases resulting from excess cytokine production.

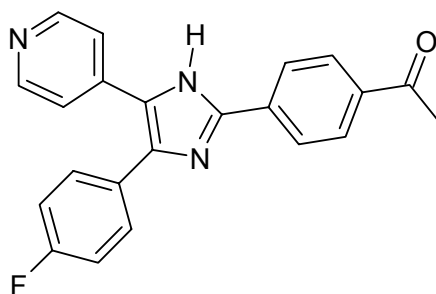
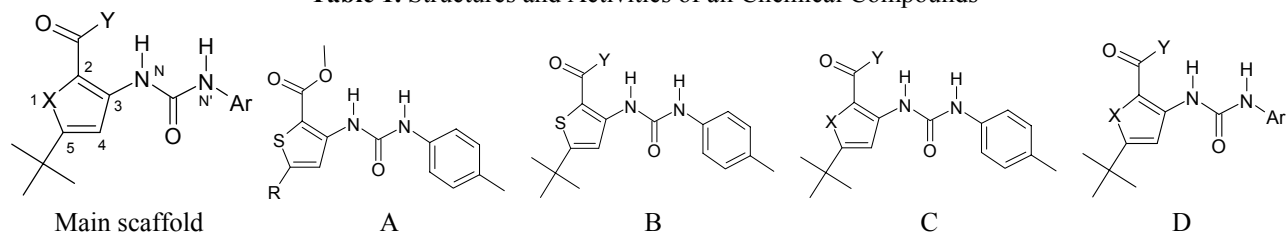


Figure 1. SB203580, an inhibitor of p38 MAP kinase.

A QSAR study for the above mentioned series of compounds was used to rationalize structural requirements for increasing binding affinity of the compounds to p38 MAP kinase. A QSAR study sought to explain and predict activities of series of congeners by utilizing empirical descriptors. QSAR modeling results in a quantitative correlation between chemical structure and biological activity [8]. Conventional QSAR relies on descriptors to characterize structures [9,10]. The factors contributing to the biological activity can be understood through use of different physicochemical descriptors in the generation of QSAR models [11,12]. Further, QSAR enables to establish *in silico* quantitative models to predict the activity of novel molecules prior to their synthesis and simultaneously provide deeper insight into the mechanism of drug–receptor interaction [13,14].

In the present study, we have gone beyond the mentioned studies to develop 3D–QSAR models, MFA [15,16,17] on a series of urea derivatives in order to provide key structural features required to design p38 MAP kinase inhibitors to efficiently predict the biological activity of compounds not included in the training set.

Table 1. Structures and Activities of all Chemical Compounds



Comp.	Scaffold	R	Y	X	Ar	Experimental	
						IC ₅₀ (nM)	pIC ₅₀
1	A	C(CH ₃) ₃	–	–	–	413	6.38
2	B	–	OEt	–	–	3020	5.51
3	B	–	OPr	–	–	482	6.31
4	B	–	O(CH ₂) ₂ OH	–	–	57	7.24
5	B	–	O(CH ₂) ₃ OH	–	–	56	7.45
6	B	–	O(CH ₂) ₂ OCH ₃	–	–	464	6.33
7	B	–	OCH ₂ CO ₂ CH ₃	–	–	5310	5.27
8	C	–	NHMe	S	–	34	7.46
9	C	–	OMe	O	–	73	7.13
10	C	–	NHMe	O	–	32	7.40
11	C	–	OMe	NH	–	33	7.48
12	D	–	OMe	S	4-C ₂ H ₅ -Ph ^a	660	6.18
13	D	–	NHMe	S	4-C ₂ H ₅ -Ph	119	6.92
14	D	–	NHMe	S	4- _i Pr ^b	270	6.56
15	D	–	OMe	S	4-F-Ph	830	6.08
16	D	–	NHMe	S	4-F-Ph	88	7.05
17	D	–	OMe	S	4-NH ₂ -Ph	750	6.12
18	D	–	OMe	S	4-OH-Ph	610	6.21
19	D	–	OMe	S	2,3-Cl-Ph	180	6.73
20	D	–	OMe	O	4-F-Ph	210	6.67
21	D	–	OMe	NH	4-F-Ph	42	7.37
22	D	–	OMe	NH	2-Cl-Ph	60	7.22
23	D	–	OMe	NH	3-Cl-Ph	27	7.56
24	D	–	OMe	NH	4-Cl-Ph	43	7.35
25	D	–	OMe	NH	2,3-Cl-Ph	6	8.22
26	D	–	NHMe	NH	2,3-Cl-Ph	44	7.35
27	D	–	OMe	NH	1-naphthyl	12	7.92
28	D	–	NHMe	NH	1-naphthyl	28	7.55
29	D	–	OMe	NMe	Ph	947	6.02
30	D	–	OMe	NMe	4-F-Ph	663	6.17
31	D	–	OMe	NMe	2,3-Cl-Ph	387	6.41
32	D	–	OMe	NMe	1-naphthyl	253	6.58
33	D	–	OMe	O	3,4-Cl-Ph	1200	5.92
34	B	–	O _i Pr	–	–	741	6.13
35	A	NH ₂	–	–	–	441	6.35
36	D	–	OMe	NMe	4-CH ₃ -Ph	400	6.30
37	D	–	OMe	S	Ph	290	6.53
38	D	–	OMe	S	4-Cl-Ph	220	6.65
39	C	–	NHMe	NH	–	67	7.17
40	D	–	OMe	NH	Ph	44	7.35
41	D	–	OMe	O	2,3-Cl-Ph	32	7.48

^a Ph = Phenyl

^b _iPr = Isopropyl

2 MATERIALS AND METHODS

2.1 Experimental Section

A data set of 41 compounds reported as p38 α MAP kinase inhibitors was collected from the literature [18]. The inhibitory activities were expressed as IC₅₀ [19] (concentration of the compound required to inhibit *E. coli* p38 α MAP kinase activity by 50%). These IC₅₀ values were further converted into the corresponding pIC₅₀ (–log IC₅₀) values (Table 1). All the compounds were built using the co–ordinates of the ligand present in crystal structure of p38 MAP kinase (PDB ID 1W82 [20]). The molecules were later minimized using steepest descent algorithm using a gradient convergence value of 0.001 kcal/mol. Partial atomic charges were calculated using the Gasteiger method [21]. Furthermore, optimization of the compound geometry was done using the MOPAC 6 package using the semi–empirical AM1 Hamiltonian [22].

2.2 Alignment

All the molecules were aligned using the shape reference alignment option [23] in the QSAR module within Cerius² [24]. The most active compound **25** was used as shape reference compound for superposing the rest of the molecules. The aligned molecules are shown in Figure 3. MFA studies were performed using QSAR module of Cerius². The negative logarithm of the biological activities of all 32 molecules in the training set and 9 molecules in the test set were chosen as the dependent variables (Tables 3 and 4). By using G/PLS regression method, multiple QSAR equations were generated with over 50,000 generations from which one equation with the highest r^2 , r^2_{cv} and lowest PRESS value was considered for further discussion. Cross–validation was done with the leave–one–out procedure. PLS analysis was scaled, with all variables normalized to a variance of 1.0.

3 RESULTS AND DISCUSSION

The training set consisted of 32 compounds while the model was validated using an external set of 9 compounds. The statistical details of the 3D–QSAR model are shown in Eq. (1). Tables 3 and 4 show the actual and predicted activities obtained from MFA 3D–QSAR models for the training and the test set molecules. Figure 2 shows the graph of actual versus predicted pIC₅₀ values of the training set and the test set molecules for 3D–QSAR models. The cross–validated r^2_{cv} for MFA model was 0.799, while the non–cross–validated r^2 with five components was 0.901. The bootstrapping r^2_{bs} value was 0.898. The actual and predicted pIC₅₀ values of the training set are shown in Table 2. The CH₃ probe represents steric and the H⁺ probe represents electrostatic descriptors in the MFA QSAR Eq. (1).

$$\begin{aligned} \text{Activity} &= 3.6421 + 0.074492 * \text{“CH}_3/465\text{”} + 0.033107 * \text{“H}^+/397\text{”} - 0.034672 * \text{“CH}_3/342\text{”} + 0.016262 * \\ &\text{“H}^+/369\text{”} + 0.102862 * \text{“CH}_3/359\text{”} + 0.016145 * \text{“H}^+/422\text{”} + 0.063181 * \text{“CH}_3/89\text{”} - 0.00722 * \text{“H}^+/289\text{”} \quad (1) \\ \text{PRESS} &= 2.664 \quad r^2_{cv} = 0.799 \quad r^2 = 0.901 \quad r^2_{bs} = 0.898 \quad r^2_{pred} = 0.740 \end{aligned}$$

where: PRESS, predicted sum of squared residuals; r^2_{cv} , cross-validated correlation coefficient; r^2 , conventional correlation coefficient; r^2_{bs} , bootstrap correlation coefficient; r^2_{pred} , predictive correlation coefficient.

The presence of steric descriptor (CH₃/89) with positive coefficient at C⁵ position of five member heterocyclic groups explains that, a steric group at this position increases the activity. This is evident in the compounds **1** and **35**. Compound **1** contains a tertiary butyl group at C⁵ where as in the compound **35** it is replaced with NH₂, which is responsible for decrease in activity. At the C² position of the heterocyclic ring, the presence of steric (CH₃/342) descriptor with negative coefficient and electrostatic (H⁺/397) descriptor with positive coefficient indicates that electron withdrawing substituent with less steric environment is favored. It shows that any substituent with O or N will have high activity may be due to hydrogen bond interaction between the substituent and active site amino acids of protein. In compounds **4–6**, when –O(CH₂)₂OH group is replaced with –O(CH₂)₂OCH₃, activity substantially decreases. At the N¹ position of the urea group, the appearance of two steric groups (CH₃/465 and CH₃/359) with positive coefficient indicates that bigger steric groups are favorable at this position. Compounds **27** and **28** and which contains naphthalene group at this position shows higher activity when compared to the compound **29**, which has phenyl group. The presence of two electrostatic (H⁺/369) descriptors with positive coefficient explains the need of both bulky and electron withdrawing substituent. Substitution of the only phenyl group at this position is having less activity as with the compound **29**. But any electron withdrawing group is substituted in phenyl ring shows higher activity (compounds **12–26**). Compound **25** with two Cl on the phenyl ring has highest activity.

Compound **36** was found to be an outlier in the test set of MFA model. Its activity is over predicted. This might be due to the presence of –CH₂Ph group on the N¹ position of the urea group. As described above bulkier groups are favored at this position it shows higher activity, but the actual activity of this molecule is less may be due to the steric clash between methyl of N–CH₃ and tertiary butyl group present at 1st and 5th position of pyridine ring respectively.

A high r^2_{cv} alone, however, is not a sufficient criterion for a QSAR model to be robust and highly predictive. The predictive power of the model was therefore validated with the test set molecules. The predictive power of the model generated was calculated with the formula $r^2_{pred} = (SD - PRESS)/SD$, where SD is the sum of the squared deviations between the biological activities of each molecule and the mean activity of the training set of molecules and PRESS is the sum of squared deviations between the predicted and actual activity values for every molecule in the test set. The prediction of the model was reasonably good with a predictive r^2 (r^2_{pred}) value of 0.740. The predictive r^2 ($r^2_{pred} = 0.983$) of MFA model was found to be high except for compound **36**.

Table 2. Actual and predicted activities of the training set molecules

Comp.	$pIC_{50\text{exp}}$	$pIC_{50\text{MFApre}}$
1	6.38	6.645
2	5.51	5.834
3	6.31	6.262
4	7.24	7.279
5	7.45	6.646
6	6.33	6.328
7	5.27	5.166
8	7.46	7.265
9	7.13	7.108
10	7.40	7.443
11	7.48	7.296
12	6.18	6.555
13	6.92	7.260
14	6.56	6.549
15	6.08	6.257
16	7.05	6.891
17	6.12	6.347
18	6.21	6.492
19	6.73	6.956
20	6.67	6.777
21	7.37	7.329
22	7.22	6.973
23	7.56	6.991
24	7.35	7.116
25	8.22	7.894
26	7.35	7.621
27	7.92	8.377
28	7.55	7.186
29	6.02	7.130
30	6.17	7.197
31	6.41	6.369
32	6.58	8.178

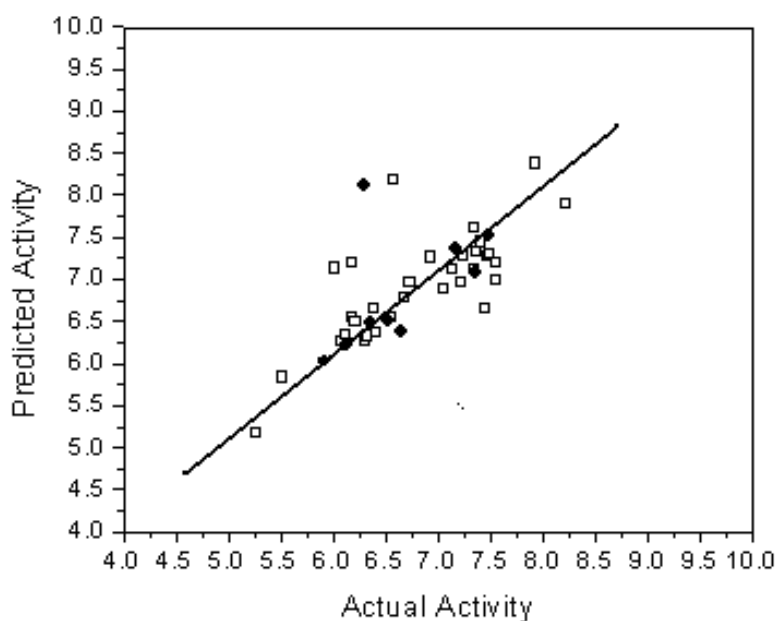


Figure 2. Scatter plots of actual versus predicted activities of Training (□) and test (■) molecules.

Table 3. Actual and predicted activities of the test set molecules

Comp.	pIC ₅₀ _{exp}	pIC ₅₀ _{MFA pre}
33	5.92	6.029
34	6.13	6.219
35	6.35	6.481
36	6.30	8.115
37	6.53	6.512
38	6.65	6.377
39	7.17	7.363
40	7.35	7.083
41	7.48	7.520

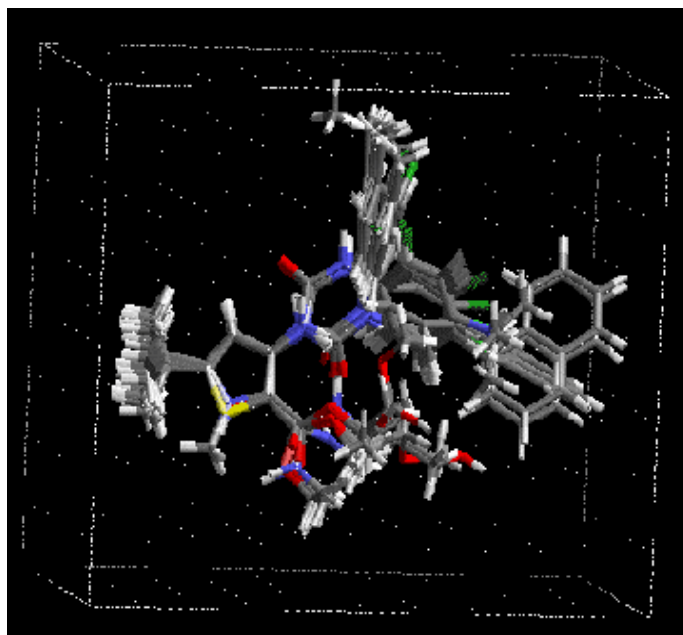


Figure 3. Alignment of study molecules in grid.

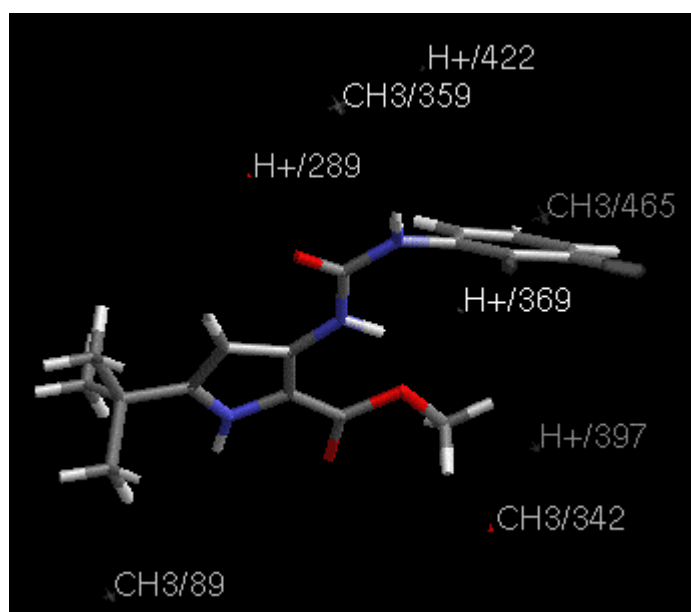


Figure 4. Mapping of 3D-descriptor points in the QSAR equations with most active molecule.

4 CONCLUSIONS

Congeneric data set of 32 compounds as selective p38 MAPK inhibitors was used to generate and validate 3D–QSAR model using MFA methodology. The best predictive model was generated with r^2_{pred} of 0.740 and had the best predictive capacity with a high LOO cross–validated correlation coefficient ($r^2_{\text{cv}} = 0.799$). It was able to predict the inhibitory activities of 9 inhibitors of the p38 MAPK not included in the model construction and offers important structural insight for rational designing novel selective p38 MAPK inhibitors prior to their synthesis. We conclude that the structural features indicated by this 3D–QSAR analysis may provide a helpful guideline to design novel compounds with inhibitory activities against p38 MAPK.

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